

MARKOV, M.; TOSHEV, Iv.

On postoperative evisceration. Khirurgiia (Soffia) 18 nc.4:
459-465 165

Okruzhen onkologichen dispanser, Varna (gl. lokar M. Markov).

DASKALOV, D.; MARKOV,M.; MINEV, Ts.

Neurovegetative reaction in women and men. Izv biol med BAN 3 no.4:
91-97 '60.

1. Institut po fiziologiia pri BAN (Direktor: akad. D.Orakhovets)
(NERVOUS SYSTEM)

MARKOV, M.

Changes in the electroconductivity of skin in contact with anod. Doklady BAN 14 no.4:413-415 '61.

1. Physiologisches Institut an der Bulgarischen Akademie der Wissenschaften. Vorgelegt von Akademiemitglied D. Orahovats [D. Orakhovats].

MARKOV, M.

Relative dependency of electroconductivity from number of active perspiratory glands after electrophoresis of acetylcholin adrenalin and pilocarpin. Doklady BAN 14 no.5:555-558 61.

1. Physiologischches Institut der Bulgarischen Akademie der Wissen - schaften. Vorgelegt von Akademiemitglied D. Orachovats[D. Orakhovats]

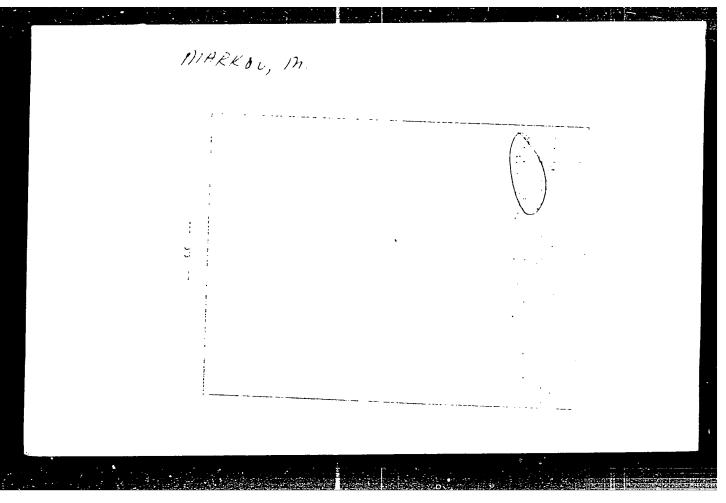
(Electric conductivity) (Cataphoresis)
(Adrenalin)

MARKOV, M.; PETROV, I.

Investigation of perspiration suggested during hypnotic condition. Doklady BAN 14 no.7:771-774 *61.

1. Vorgelegt von Akademiemitglied D. Orahovats [D. Orakhovats].

(PERSPIRATION) (HYPNOTISM)



MARKOV, M.; PETROV, I.

On the mechanism of gustatory sweating. Dokl. Bolg. akad. nauk 15 no.1:89-92 '62.

1. Vorgelegt von Akademiemitglied D. Orahovats. (SWEATING physiol) (TASTE physiol)

GRAHOVATS, D.; MARKOV, N.; WARBANOV, W.

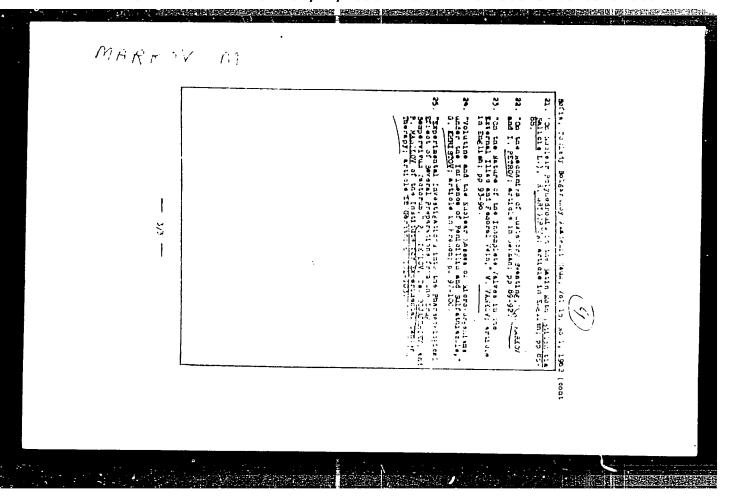
Effect of the resistance of the soft tissue of the extremities in measuring blood pressure with Korotkoff's method. Dokl. Bolg. akad. nauk 15 no.3:333-336 62.

(HLOOD PRESSURE) (LEG blood supply)

MARKOV, M.

Changes in the electrical conductivity and potential of the skin in pressure on the anode in relation to the initial degree of electrical conductivity and potential differences. Izv. inst. fiziol. 5:179-186 162.

(SKIN physiol)



ORAHOVATS, D. [Orakhovats, D.], akad.; MARKOV, M.; WARBANOV, W. [Vurbanov, V.]

Effect of the resistance of soft tissues in the measurement of blood pressure by the Korotkov method. Doklady BAN 15 no.3:333-336 162.

1. Physiologisches Institut der Bulgarischen Akademie der Wissenschaften. 2. Chlen Redaktsionnoy kollegii, "Doklady Bolgarskoy Akademii Nauk" (for Orakhovats).

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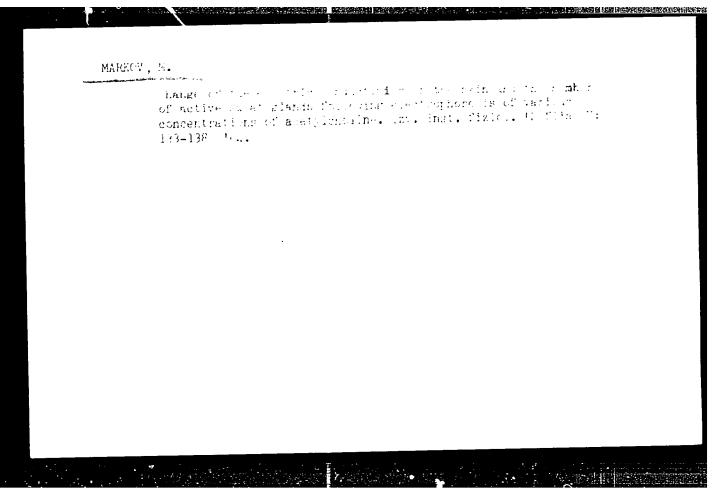
ORAKHOVATS, D. [deceased]: DRAGANOV. 1.; VORMANOV. V.: BARKOV, K.: FIRVY, A.:

FAMCHOV. 7.; NATURY, Th.

Simultaneous recording of the anternal pressure, soft orrathly,
Korotkovia tones and pressure of the outful the tracils, artern
by means of direct measurements in san. fov. than. first. 1.0ff.c.
7:5-18 1.4.

ORAKHOVAIS, D. [deceased]; MARKOV, M.; VIRBANOV, V.; PRAGANOV, L.

Studies on the development of Morotkov's tones. Izv. Inst. fizi ..
(Sofiia) 7:19-28 '64.



ACC NR: AP7008901

SOURCE CODE: UR/0386/66/003/002/0098/0100

AUTHOR: Markov, M.

ORG: Physics Institute imeni P. N. Lebedev, Academy of Sciences USSR (Fizieneskiy Institut AN USSR)

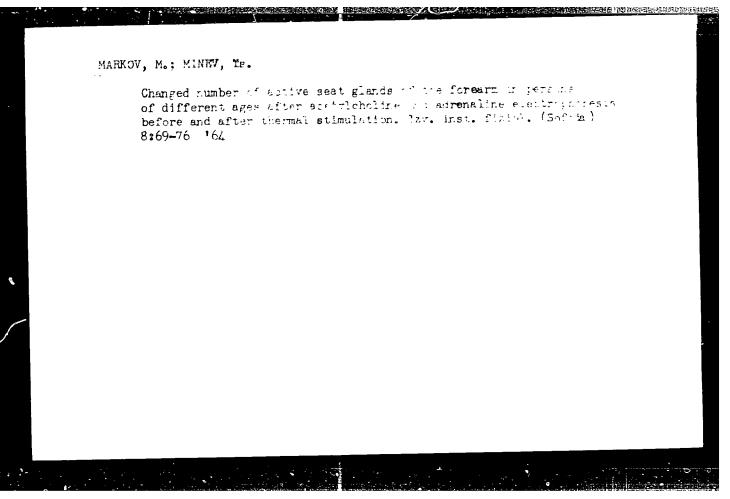
TITLE: Search for specific mu-meson and sup nu mu-neutrino interactions at very mag.

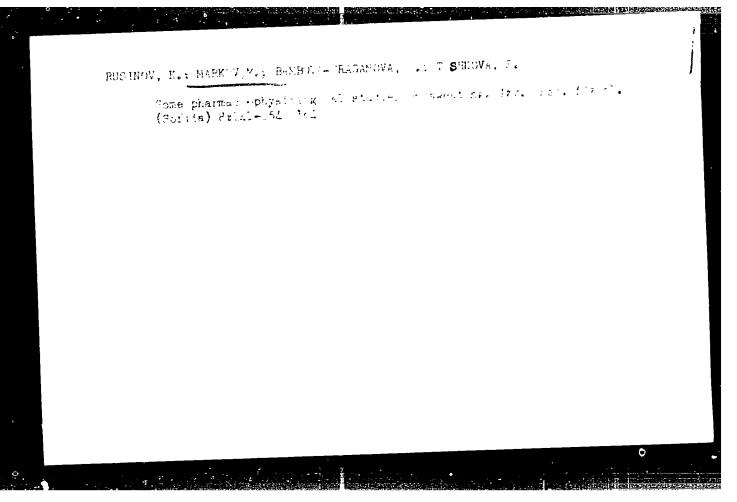
COURCE: Unurual eksperimental*noy i teoreticheskoy fiziki. Pis*ma v rejaktslym, v. j, no. 2, 1966, 98-100

TWPIC TABS: mu meson, meson interaction, neutrino, cosmic ray SUB SUB: 20

AN TRACT: New information mathered from cosmic ray studies seems to indicate that at very high energies there may exist specific \mathcal{A} -meson and \mathcal{A} -meson and interactions in addition to the usual electromagnetic and weak interactions. The author investigates a pseudovector (pseudomagnetic and weak interactions. If $g(\mathbb{A}/\mathbb{A}) = \mathbb{A} = \mathbb{A}$ and $g(\mathbb{A}/\mathbb{A}) = \mathbb{A} = \mathbb{A}$ interactions. In the first approximation the g-charge is carried by a \mathbb{A} -meson and nucleons only; a \mathbb{A} -neutrino and electron carry a much weaker g-charge. An evaluation of \mathbb{A} -pair production by pseudophotons in Coulomb fields of extended nuclei is given for $\mathbb{A} = \mathbb{A} = \mathbb{A}$

Cord 1/1





MARKOV,M.

Age changes in the electrical conductivity of the skin after the alectrophoretic administration of acetylcholine, adrenaline and pilocarpine. Dokl. Bolg. akad. nauk 18 no.22183-186 '65

1. Submitted on October 16, 1964

5(3) 5.5600

66965

AUTHORS:

Kondrat'yev, D. A., Markov, M. A., Minachev, Kh. M.

501/32-25-11-13/69

作。1965年,1966年,1965年,1966年,1966年,1966年,1966年,1966年,1966年,1966年,1966年,1966年,1966年, 1966年 - 1966年

TITLE:

Analysis of Mixture of C, to C, Hydrocarbons by the Method of

Liquid - Gas Chromatography

PERIODICAL:

Zavodskaya laboratoriya, 1959, Vol 25, Nr 11, pp 1301-1304

(USSR)

ABSTRACT:

A simple device with a microflame detector (Ref 1) designed for the analysis of paraffin hydrocarbons, naphthenes, aromatic, and some unsaturated C, to C, hydrocarbons has been developed.

The construction of the dosing evaporator and the microflame

detector was based on a device developed by B. A. Rudenko

(Izvestiya AN SSSR, in the press). The detector is , in principle, a burner consisting of a capillary 1 mm in diameter. The burner is located at the upper output of a column 6 m long, and is connected to a Chromel-Alumel thermocouple (Fig 1: Scheme of the device). Data obtained with the detector are recorded by

automatic recording potentiometer of the types PS1-01 \$\mathcal{P}\$ or EPP-09, with a second PP potentiometer connected thereto. The hydrocarbons may be analyzed by means of a thermostat of the

Card 1/3

APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R001032510002-7"

Analysis of Mixtures of C₅ to C₇ Hydrocarbons by the Method of Liquid - Gas Chromatography

SOV/32-.5-11-13/6)

type $\overline{\text{TS-15M}}^{16}$ at constant or variable temperature. The best separating efficiency was reached when two columns (each 3 m long) were used, the one filled with diatomite brick chips (0.25 to 0.5mm) and tricresyl phosphate, and the other filled with diatomite brick chips and dioctyl phthalate. Separation was first effected at 15 - 20°C (for 15 minutes), and all normal and isoparaffin hydrocarbons C_5 to C_7 were separated

normal and isoparaffin hydrocarbons C_5 to C_7 were separated from one another, whereafter temperature was raised to 85° C (1.5° per minute). Hydrogen was passed through the system with a rate of 60 cm³ per minute. The chromatogram of a 15-component (C_5 to C_7 hydrocarbon) mixture shows that all substances could

be separated except for the pairs 2,3-dimethylbutane-2-methylpentane, cyclopentane-3-methylpentane, and cyclohexane-3-methylhexane. Results of an analysis of an artificial hydrocarbon
mixture (Table 1) as well as with catalyzates at elevated
temperature and hydrogen pressure (Table 2) are given. There
are 3 figures, 2 tables, and 2 Soviet references.

Card 2/3

Analysis of Mixtures of C₅ to C₇ Hydrocarbons by the Method of Liquid - Gas Chromatography

SOV/32-25-11-13/69

ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo Akademii nauk SSSR (Institute of Organic Chemistry imeni N.D.Zelinskiy of the Academy of Sciences of the USSR)

Card 3/3

MINACHEV, Kh.M.; SHUYKIN, N.I.; MARKOV, M.A.

Effect of the specific surface of a platinized aluminosilicate on the degree of n-nonane conversion. Report No.1: Change in the activity of platinized aluminosilicate in the course of the treatment of the carrier with hydrogen. Izv.AN SSSR Otd.khim. nauk no.5:907-912 My 160. (NIRA 13:6)

1. Institut organicheskoy khimii imeni N.D. Zelinskogo Akademii nauk SSSR.

(Aluminosilicates) (Platinum) (Nonane)

s/062/60/000/008/023/033/XX BO13/BO55

AUTHORS:

Minachev Kh. M., Shuykin, N. I., and Markov, M. A.

TITLE:

Investigation of the Effect of the Specific Surface of Platinized Alumosilicate on the Degree of n-Nonane

Conversion

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk.

1960. No. 8, pp. *466 1470

TEXT: This paper is a continuation of the studies on the effect of the specific surface of catalysts on the degree of hydrocarbon conversion. For this purpose, the authors studied the conversion of n-nonane on alumo. silicate catalysts with widely varying specific surfaces. Two 0.5% platinum/alumosilicate catalysts with specific surfaces of 320 m^2/g (K-1) and 60 m2/g (K-?) were used. A standard alumosilicate catalyst for the cracking process was used as carrier for the preparation of the platinum catalysts. To prepare the latter, the carrier was saturated with a dilute H2PtCl6 solution. The data of the n-nonane used were in agreement with those given in Ref. 4. Infrared spectra showed the n-nonane to be free Card 1/3

S/062/60/000/008/023/033/XX Investigation of the Effect of the Specific Surface of Platinized Alumosilicate on the B013/B055 Degree of n-Nonane Conversion

of isomers. The experimental apparatus has been described in Ref. 1. The experiments were carried out in a continuous system at 360 to 450°C. a hydrogen pressure of 10 atm and a flow rate of 1 h-1. The molar ratio of hydrocarbon and hydrogen was 1:5. A fresh catalyst was used for each experiment. The results of the examination of the catalyzates are listed in Tables 1 and 2 and graphically represented in Figs. 1 - 3. These data show that the degree of hydrocarking of hydrocarbons considerably decreases with decreasing specific catalyst surface. This is in agreement with data given in Ref. 2. The yields of hydrocracking products on K-1 catalysts were found to increase more rapidly with a temperature rise than on K-2 catalysts. At temperatures of 420° - 450°C, hydrocracking on K-2 is insignificant, which enables c_q isoalkanes to be obtained in comparatively high yields (54% at 450°C). On K 1, the maximum yield of isononanes is obtained at 380°C (53%). Since arcmatization occurs to a noticeable degree only at 400°C, aromatic hydrocarbons can be obtained over K-2 before Co isoalkane yields are reduced. The total yield of hydrogracking products under the experimental conditions was 45.7% over K-1, and 25.3% over K-2. The experiments have thus shown that by decreasing the specific surface Card 2/3

Investigation of the Effect of the Specific S/062/60/000/008/023/033/XX Surface of Platinized Alumosilicate on the Degree of n-Nonane Conversion S/062/60/000/008/023/033/XX

of the catalyst the process can be carried out at higher temperatures without the occurrence of hydrographing. There are 3 figures, 2 tables, and 4 Soviet references.

ASSOCIATION: Institut organicheskey khimii im. N. D. Zelinskogo Akademii

nauk SSSR

(Institute of Organic Chemistry imeni N. D. Zelinskiy of

the Academy of Sciences USSR;

SUBMITTED: February '8, 1959

Card 3/3

50V/7/1-30-3- 1/1/1

AUTHORS:

5.302

Tits-Skvortsova, I. N., Danllova, T. A., Markov, M. A.,

Stepanova, I. I., Osipenko, Ts. D.

TITLE:

Synthesis and Conversions of Sulfur Compounds of

Naphthalene Series Over an Alumina-Silica Catalyst

PERIODICAL:

Zhurnal obshchey khimii, 1960, Vol 30, Nr 3, pp 985-

991 (USSR)

ABSTRACT:

The following compounds were synthesized and their conversions over an alumina-silica catalyst at 300 was studied. a - Thionaphtnol (72%), bp 143-1440

(6 mm); \mathcal{B} -thionaphthol (80%), mp 79-80°; α -napthyl decyl sulfide (72%); α -naphthyl cyclopentyl sulfide (45.6%), bp 168-168.5° (2 mm), n_D^{20} 1.6419, d_4^{20} 1.1193;

 \mathcal{B} -naphthyl decyl sulfide (68%), bp 209-219° (2.5 mm), mp 34-35°; \mathcal{B} -naphthyl cyclopentyl sulfide (65%),

Card 1/5

bp $187.5-188^{\circ}$ (4 mm), n_D^{20} 1.6455, d_4^{20} 1.1052. This

Synthesis and Conversions of Sulfur Compounds of Naphthalene Series Over an Alumina-Silica Catalyst

78297 **SOV/**79-30-3-51/69

study was undertaken to see whether the conversions of the thionaphthols over the above catalyst at 300° proceed similarly to the conversions of arcmatic thiols under the same conditions. Conversions of aromatic thiols proceed as authors showed (DAN SSSR, 80, 377, 1951; ZhOKh, 21, 212, (1951); and others), according to the following scheme:

$$\begin{array}{c} SH \\ \longrightarrow \\ R \end{array} + H_2S \tag{1}$$

It was found that both α - and β -intonaphthols undergo an identical conversion over this catalyst at 300°, according to the following scheme:

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Synthesis and Conversions of Sulfar Commendation of Naphthalene Series Over an Alumina-Silina 307, fres0-3-51/200 Catalyst

Comparison of schemes 1 and 2 shows that the isomeric lpha - and eta -thionaphthols and aromatic thiols undergosimilar conversions over the same catalyst at the same temperature. &-Naphthyl decyl sulfide decomposes over the catalyst at 300° to form naphthalene (36%, of weight of catalyst), decyl mercaptan (13.1%), decen (7.8%), and H₂S, according to scheme:

$$S = C_{10}H_{21}$$

$$+ C_{10}H_{21}SH$$
(3)

α-Naphthyl cyclopentyl sulfide decomposes over the catalyst to form naphtnalene (40% of weight of catalyst), cyclopentanthiol (6.6%), dicyclopentyl sulfide (2.2%) and H₂S. The reaction proceeds also analogously to scheme 3. Catalytic decomposition of **B**-naphtnyl cyclopentyl sulfide under above conditions results in the formation of **B**-thionaphthol (15.6% of weight of catalyst), cyclopentene (10.2%)

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"APPROVED FOR RELEASE: 06/14/2000

Synthesis and Conversions of Sulfur Compounds of Naphthalene Series Over an Alumina-Silica Catalyst

78297 SOV/79-30-3-51/69

naphthalene (43.5%) and ${\rm H_2S}$, according to a different scheme:

Catalytic decomposition of β -naphthyl decylence with under the same conditions results in the formulation of: β -thiomaphthol (1.1% of weight of catalyst), decylence mercaptan (6%), naphthalene (30.5%), decenter where fraction (4.2%) and H_OS, addording to:

$$S = C_{10}H_{21} + C_{10}H_{20}$$

$$S = C_{10}H_{21} + C_{10}H_{21}SH$$
(5)

Card 4/5

Synthesis and Conversions of Sulfur Compounds Synthesis and Conversions of Sulfur Compounds 78297 of Naphthalene Series Over an Alumina-Silica SOV/79-30-3-51/69 Catalyst

The comparative strength of the sulfur bond with different radicals is shown in scheme 6:

$$C_{0}H_{5} = S \stackrel{1}{+} C_{10}H_{21} \qquad C_{10}H_{21} = S \stackrel{1}{+} C_{6}H_{11}$$

$$C_{0}H_{5} = S \stackrel{1}{+} C_{5}H_{9} \qquad 3 \cdot C_{10}H_{7} = S \stackrel{1}{+} C_{1}H_{9}$$

$$C_{0}H_{5} = S \stackrel{1}{+} C_{6}H_{11} \qquad C_{10}H_{21} = S \stackrel{1}{+} 2 \cdot C_{10}H_{7}$$

$$C_{10}H_{21} = S \stackrel{1}{+} 2 \cdot C_{10}H_{7} \qquad (6)$$

There are 3 tables; and 14 references, 1 U.S., 1 Dutch, 4 German, 8 Soviet. The U.S. reference is: E. D. Rossini and others, Selected Physical Values and Thermodynamic Properties of Hydrocarbons and Related Compounds (1953).

ASSOCIATION:

Moscow State University (Moskovskiy gosudarstvennyy

universitet)

SUBMITTED:

March 5, 1959

Card 5/5

MINACHEV, Kh.M.; MARKOV, M.A.; LOGINOV, G.A.

Conversions of five- and six-membered cyclenes on rareearth oxides. Neftekhimia 1 no.3:356-361 My-Je '61. (MIRA 16:11)

1. Institut organicheskoy khimii AN SSSR imeni N.D. Zelinskogo.

5. 1190

31746 5/204/61/001/004/003/005 E075/E185

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AUTHORS:

Minachev, Kh.M., Markov, M.A., and Shchukina O.K.

TITLE

Dehydrogenation of cyclohexane on the oxides of

rare earth elements

PEFIODICAL: Neftekhimiya, v.1, no.4, 1961, 489-493

TEXT: Eight oxides of rare earth elements and yttrium oxide were used as catalysts for the dehydrogenation of cyclohexane. The catalysts were prepared by dissolving the commercial oxides in 27% nitric acid, and precipitating with 12% ammonia solution at 50-60 °C. The dried and washed precipitates were compressed into 4 x 4 mm cylinders and heated at 560 °C in dry air for 8 hours. The surface areas of the oxides so obtained were determined by benzene vapour adsorption. The dehydrogenations were carried out at 515-590 °C under atmospheric pressure. The catalysts were activated at 560 °C for 2 hours by passing through them currents of air, hydrogen or 1 hours by the catalysts with the greatest activity were produced by the activation with nitrogen. Experimental results show that all the catalysts dehydrogenate cyclohexane to benzene

Card 2/45

Dehydrogenation of cyclohexane ... $\frac{317 \mu 6}{5/204/61/001/004/003/005}$ E075/E185

The determination of specific areas of the catalysts permitted the calculation of the specific activity and specific toke formation for the various catalysts, and thus their relative overall activities could be compared. The results are given in Table 4. It can be seen that the specific activity and coke formation at 530-560 °C does not differ much inside the yttrium group of the oxides. The yttrium group oxides exceed the cerium oxide group in respect of activity. It was shown that there exists linear dependence between the logarithms of the percentage conversion and the reciprocal temperature of the reaction. The energies of activation calculated from the slopes of the lines had typical values for acidic catalysts in the case of neodymium, gadolinium and holmium oxides, but exceeded 50 ktal for the remaining oxides. There are 4 figures, 4 tables and 8 references: 5 Soviet bloc and 3 non-Soviet-bloc. The English language references read. Ref. 1: R.A Briggs, H.S. Taylor, J. Amer. Chem. Soc., v.63, 2500, 1941.
Ref. 4; V.I. Komarevsky, Ind. Eng. Chem., v 49, 264, 1957.

Dehydrogenation of cyclohexane ... 5/204/61/001/004/003/005 E075/E185

ASSOCIATION: Institut organicheskoy khimii AN SSSR im.

N.D. Zelinskogo
(Institute of Organic Chemistry AS USSR imen: N.D. Zelinskiy)

SUBMITTED: June 7, 1961

Card 3/4

MINACHEV, Kh.M.; MARKOV, M.A.; SHCHUKINA, O.K.

Dehdrocyclization of n-heptane over rare earth oxides. Neftekhimiia 1 no.5:610-612 S-0 '61. (MTRA 15:2)

1. Institut organicheskoy khimii AN SSSR imeni N.D.Zelinskogo. (Heptane) (Aromatization) (Rare earth oxides)

s/081/62/000/009/032/075 8158/8101

等。1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年,1995年

AUTHORS: Tits-shvortsova, I. N., Danilova, T. A., Larkov, A. A., Stephnova, I. I., Osipenko, Ts. D.

TITLE: Conversion of organosulfur compounds of the α - and β -naphthalene series in the presence of an aluminosilicate catalyst

P.RIODICAL: Referetivny zhurnal. Khi iya, no. 9, 1962, 228, abstract 77h180 (Jb. "Khimiya seraorgan. soyedineniy, soderzhezhokikhou, v neftyckh i nefteproduktakh. v. 4", L., Gostoptekhimaat, 1761, 141 - 144)

TEXT: Contact conversions of organosulfur compounds of naphthalene as carried out at 300°C on an aluminosilicate catalyst under conditions described earlier (Zh. obshch. khimiya, v. 21, 1951, 242) are reexamined. In and 3-thionaphthols (4- and 4-II) were synthesized for research, 3- and 3-thionaphthyldecylsulfides (4- and 4-II) and 3- and 3-naphthylcyclopentyl-sulfides (4- and 4-III) synthesized for the first time. It was found that under these conditions 3-I and 3-I are converted to C10H8 and H2S similarly under these conditions 3-I and 3-I are converted to C10H8 and H2S similarly to the thiophenols studied earlier the respective yields being 52 and 43.00 Card 1/2

S/081/62/000/009/032/0-5
Conversion of organosulfur compounds ... B158/B101

by weight of catalyst. As established previously (see UCh, zap. 190, v.101, 1953, 263), in the case of mixed sulfides of the C_6H_5SR type (R being an alkyl or cycloalkyl), the bond between the sulfur and R is always ruptured. In the case of $^{\circ}$ -II, it was found that $C_{10}^{\rm H}_{8}$ and $C_{10}^{\rm H}_{21}^{\rm SH}$ are formed with further conversion of the latter to $c_{10}^{\rm H}_{\rm 2C}$ and $c_{20}^{\rm H}_{\rm 2C}$ and $c_{20}^{\rm H}_{\rm 2C}$ and $c_{20}^{\rm H}_{\rm 2C}$ i. the same way, forming C 10 8 and cyclopentanethiol with subsequent conversion of the latter to dicyclopentylsulfide and ${\rm H_2S.}$ % -III under these consitions decomposes to 7-1, syclopentone, $c_{10}^{\rm H}_{8}$ and $c_{20}^{\rm H}_{20}$. In the case of 4-II, -I, Contact and Hos were detected. sequently the bond between the sulfur and the benzene ring in liked milfides is much more stable and was not ruptured in any of the cases examine . The bond between the sulfur and the $c_{10}^{\rm H}8$ in the .-position is far less stable. The bond between the sulfur and the alkyl and naphthyl in the -position is more stable than that between the sulfur and naphthene rings. [Abstracter's note: Complete translation.] Card 2/2

MINACHEV, Kh.M.; MARKOV, M.A.; KHODAKOV, Yu.S.

Effect of gamma rays on the catalytic activity of platinized aluminosilicate. Izv. AN SSSR. Otd.khim.nauk no.7:1227-1230 Jl '61. (MIRA 14:7)

1. Institut organicheskoy khimii im. N.D. Zelinskogo AN SSSR. (Aluminosilicates) (Catalysis) (Gamma rays)

S, 077, 11 000 00A, 00A, 010 B1173B201

A STATE OF THE STA

AUTHORS: Minachev, Kh. M., Markev, M. A., and St. Cakerer, D. K.

TITLE: Investigation of the catalytic properties it rare earths

PERIODICAL: Akademiya nauk SSSR. Izvestaya. Cinete, the Brook backing

nauk, no. 8, 1961, 1507.1511

TEXT: The authors investigated the caracyi. It is and a mixed catalyst from 15 % erroum vide and it is specific surface of the catalysts are determined by A. M. Aubinshieyn and V. A. Arabida An SSSR. Otd. khim. n., 1955. 1296) amounts and the catalysts are marked by a caracter of the properties were marked as a market of the properties.

listed in Table 1. The experiments were made to temperatures from 5450-590°C and atmosphere in the current. The volume rate of the supplied substance of the s

0.25 hr. Before each experiment the distribution with hydrogen at 550°C. After the experiment the distribution with hydrogen at 550°C.

APPROVED FOR RELEASE: 06/14/2000 CIA

CIA-RDP86-00513R001032510002-7"

Investigation of the catalytic... Fig. 3.5.

5000-52000 with air, which was diluted with the same of 5-7%. During the regeneration the same of catalyst was determined. The experiment of the same of the same

Card 2/6

Investigation of the catalytic

B/062/61 000/006, 006/010 B117/B204

hydrocarbons, the yield of liquid catalysates on pure $\mathrm{Er}_{2}\mathrm{O}_{3}$ was much higher than on $\mathrm{Er_20_3/Al_20_3}$. However, due to the lewer stability of n-heptane and ethyl cyclopentane it was not so big as for hypoletexame. Pure Er₂O₃ thus has a much weaker cracking effect than $\text{Er}_2\text{C}_3/\text{Al}_2\text{O}_2$. The analysis of gaseous catalysis products showed that at any rate the gas obtained on $\mathrm{Er_20_3/Al_20_3}$ is richer in hydrocarbons than that thered on $\mathrm{Er}_2\mathrm{O}_3$. During the conversion of cyclohexane and nevertage the entant of saturated and unsaturated hydrocarbon is about equal In the attitions of ethyl cyclopentane, the gas produced on $\mathrm{Er}_{\mathcal{I}} C_{\mathcal{I}}/A_{-\mathcal{I}} C_{\mathcal{I}}$ and \mathcal{I} unsaturated and 10.4 % saturated hydrocartons. The analysin of the liquid catalysis products showed that the product obtained through the intersion of cyclohexane consists of unchanged hydromarbon, bertere mathal of lapentane and cyclohexane. The benzene content in the profit is traced on Er203/Al203 amounted to 4.1 % at 54500 and 20.4 % at 5000 content. $\mathrm{Er}_2\mathrm{O}_3$, the benzene content at 545°C was 1.7 β and at 595°C true of . The Card 3/6

Investigation of the catalytic ...

Card 4/6

unchanged n-heptane, toluene and benzene was prestified tonic conversion of n-heptane. The content of aromatic ngarocarters was wrong equal in the liquid catalysate in the presence of both ratalysts. The yield of liquid catalysate, however, was much higher on pure Er C, than on Er₂0₃/Al₂0₃. More aromatic hydrocarnons are thus illusined Er.O. The presence of becomes in the per hydrocarbon used than on Er₂O₃. catalysis products of n-heptane points towards the demetrylation process. When passing through ethyl cyclopentane, no loldere was established or Er203. In the product obtained on Er203/Al203 however 8.150 tillene were found. The iodine numbers of the products obtained during the conversion of cyclohexane and n-heptane on both catalysts were squally not higher than 15. The iodine number of the product of the ned from J lehexane on the mixed catalyst at 590°C, i.e. 29, was an ar exite. During catalysis of ethyl cyclopentane, the icdine numbers of the adulysates were 29 on Er203 and 52 on Er203/Al2C3. There are it wires I table. and 10 references: 6 Soviet-bloc and 4 non-Sov et 15

Investigation of the catalytic ...

S/062/61/000/008/008/010 B117/B206

references to English-language publications read as follows: R. A. Briggs, H. S. Taylor, J. Amer. Chem. Soc. 63, 2500 (1941); V. I. Komarevsky, Industr. and Engng. Chem. 49, 264 (1957); g. E. Green, Nature 180, N 4580, 295 (1957).

ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo Akademii

nauk SSSR (Institute of Organic Chemistry imeni

N. D. Zelinskiy, AS USSR)

SUBMITTED:

December 13, 1960

Card 5/6

27493 8/062/61/000/009/008/014 B117/B101

A STATE OF THE PARTY OF THE PAR

2209, 1274 1273 5 1190

Minachev, Kh. M., Markov, M. A., and Shchukina, O. K. AUTHORS:

Study of the catalytic properties of rare earth oxides TITLE:

2. Transformation of cyclohexene, 1-methyl cyclohexene-1, and

n-heptene-1 on erbium oxide

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye khimicheskikh

nauk, no. 9, 1961, 1665-1669

TEXT: The present work which continues the studies on the catalytic properties of rare earth oxides was undertaken with a view to investigating the behavior of several cycloalkenes on pure erbium oxide and a mixed catalyst (15% erbium oxide, 85% aluminum oxide) at 510°-515°C. Catalyst preparation, reaction conditions and method of analyzing the catalyst have been described previously (Ref. : Izv. AN SSSR. Otd. khim. n. 1961, no 8) The following hydrocarbons were used: cyclohexene; 1-methyl cyclohexene-1, n-heptene-1; the yields of liquid catalysis products obtained by passing the two first-mentioned cycloalkenes over Er203 were independent of time

and amounted to 80.7% and 83.4%, respectively. They contained no hydro Card 1/6

27493 \$/062/61/000/009/008/014 B117/B101

Study of the catalytic properties

carbons of molecular weight higher than that of the initial substance. Chromatographic analysis showed that the gas formed simultaneously is made up of 90-96% hydrogen and 4-10% C1 to C4 hydrocarbons. Carbonization on Er203 was insignificant. The product obtained from cyclohexene over Er203 consisted of benzene (20.23%) and cyclohexene. In experiments using 1-methyl cyclohexene-1 and the same catalyst, 1-methyl cyclohexene-1, a mixture of 1-methyl cyclohexene-2 and 1-methyl cyclohexene-3, and toluene were identified in the reaction product. The results obtained in the conversion of 1-methyl cyclohexene-1 on Er203 are given in Table 2 tests with cyclohexene and 1 methyl cyclohexene-1 over Er203/Al203 showed that the yield of liquid products increases with time. The gases analyzed consisted of 85%-93% hydrogen and 7-15% C, to C, hydrocarbons. As in the case of Er203, the gas was richer in hydrocarbons towards the beginning of Carbonization on Er203/Al203 was higher than on pure the experiments. erbium oxide. The composition of the catalyzate obtained from cyclohexene over Er203/Al203 is represented in Table 3. The composition of the Card 2/6

27493 \$/062/61/000/009/006/014 B117/B101

Study of the catalytic properties ...

Card 3,6

catalyzate obtained from 1-methyl cyclohexene-1 was only partially clarified, since the chromatogram was greatly complicated by the formation of dimethyl cyclopentenes. A sample drawn within the first two hours was found to contain 63% toluene. The total yield of catalysis products in tests with n-heptene-1 over ${\rm Er_2O_3}$ was 74.5% and over ${\rm Er_2O_3}/{\rm Al_2O_3}$ 68.3%. The gas formed over ${\rm Er_2O_3}/{\rm Al_2O_3}$ contained double the quantity of hydrocarbons that was obtained over ${\rm Er_2O_3}$. The liquid catalysis products contained 6% toluene in the case of ${\rm Er_2O_3}$ and 8.5% in the case of ${\rm Er_2O_3}/{\rm Al_2O_3}$. The chromatogram of the product obtained over ${\rm Er_2O_3}/{\rm Al_2O_3}$ exhibited six peaks in the C₇ hydrocarbons region, apart from toluene, as compared to two peaks in the case of ${\rm Er_2O_3}$. ${\rm Er_2O_3}$ catalysis yielded products containing 91.8% unsaturated hydrocarbons and ${\rm Er_2O_3}/{\rm Al_2O_3}$ 47.0%. In conclusion, the investigation of these two catalysts yielded the following results: They differ inasmuch as the mixed catalysts produced isomerization of the 6-membered ring to a 5-membered ring, whereas this isomerization of the 6-membered ring to a 5-membered ring, whereas this isomerization

Study of the catalytic properties ...

27493 **8/062/61/000/009/008/01:** B117/B101

tion does not occur with pure erbium oxide. Both catalysts dehydrogenate the tested cycloalkenes to corresponding aromatic hydrocarbons and cause shifting of the double bond in the ring. There are 3 figures, 3 tables and 5 references: 3 Soviet and 2 non-Soviet. The two references to English-language publications read as follows: F. G. Rossini, K. S Pitzer, R. L. Arnett, R. M. Braun, G. C. Pimentd, Selected values of physical and thermodynamic properties of hydrocarbons and related compounds, Carnegi Press. 1953; E. Gil-Av. J. Herling, J. Shabtai. Chem. and chem. Ind. no. 9, 1483 (1957).

ASSOCIATION: Insitut organicheskoy khimii im. N. D. Zelinskogo Akadem:

nauk SSSR (Institute of Organic Chemistry imeni N. D

Zelinskiy of the Academy of Sciences USSR)

SUBMITTED:

December 13. 1960

Card 4/6

LUKINA, M.Yu.; ZOTOVA, S.V.; MARKOV, M.A.; OVODOVA, V.A.; KAZANSKIY, F.A., akademik

Transformations of isopropenylcyclopropane in the presence of kieselguhr. Dokl. AN SSSR 139 no.2:381-384 Jl '61. (MIRA 14:7)

1. Institut organicheskoy khimii im. N.D. Zelinskogo AN SSSR. (Propene) (Kieselguhr)

8/204/62/002/002/001/007 1060/1242

AUTHORS:

Minachev, Kh.M., Markov, M.A., and Bogomolov, V.I.

TITLE:

Conversion of cyclohexane and n alkanes on rare earth

oxides

PERIODICAL: Neftekhimiya, v.2, no.2, 1962, 144-149

TEXT: This work is a continuation of the investigation of the catalytic effect of oxides of rare earth elements on dehydrogenation and dehydrocyclization of hydrocarbons. Neodymium, helmium, and yttrium oxides on activated carbon have shown a high activity in the dehydrogenation of cyclohexane and in dehydrocyclization of n alkanes. When the concentration of Nd₂O₃ in the catalyst increases from 0.5 to 5%, the output of benzene from cyclohexane at 500° grows from 19.6 to 49.4%, while higher concentrations of oxide up. to 15% do not produce

Card 1/2

S/204/62/002/002/001/007 1060/1242

Conversion of cyclohexane...

any noticeable increase of output of benzene. The gaseous products of catalysis wer analyzed in a XT-2M (KhT-2M) unit; the liquid products were analyzed by gas-liquid chromatography. There are 4 figures and 4 tables.

ASSOCIATION: Institut organicheskoy khimii AN SSSR im. N.D. Zelinskogo (Institute of Organic Chemistry im.

N.D. Zelinskiy, AS USSR)

January 25, 1961 SUBMITTED:

Card 2/2

ACCESSION NR: AT4035162

\$/0000/63/000/000/0125/0130

AUTHOR: Minachev, Kh. M.; Markov, M. A.

TITLE: Investigation of the catalytic properties of the rare-earth elements in the transformation of hydrocarbons

SOURCE: AN SSSR. Institut geokhimii i analiticheskoy khimii. Redkozemel'ny*ye elementy* (Rare-earth elements). Moscow, Izd-vo AN SSSR, 1963, 125-130

TOPIC TAGS: rare earth, catalytic dehydrogenation, rare earth oxide, cyclohexane, n-heptane, ethylcyclopentane, methyl-cyclopentene, methylcyclohexene, n-heptene, dehydrogenation

ABSTRACT: In a series of experiments on the catalytic activity of the rare earth oxides, oxides of La, Nd, Sm, Gd, Ho, Er, Tu, Yb and Y were used as catalysts in the dehydrogenation of cyclohexane, n-heptane, ethylcyclopentane, 1-methylcyclopentane, l-methylcyclopentane, l-methylcyclohexane was first investigated at 515-580C. Determination of the specific surface of the investigated oxides made it possible to compare their specific catalytic activity and specific coke-forming power during the reaction. Calculation showed that the activation energy for Nd203, Gd203 and Ho203 are close to the values of typical oxide catalysts, while for other oxides the values exceed 50 kcal/mole. Similar

ACCESSION NR: AT4035162

results were obtained with n-heptane and cycloolefins. The isomerization of 1-methylcyclopenten-I and 1-methylcyclohexen-I is plotted against temperature, and the production of benzene from the latter is compared with different catalysts. Finally, the reduction of cyclohexane and n-heptane is studied over oxides of the rare-earth elements adsorbed onto activated charcoal. Such catalysts were found to be significantly more effective than the oxides alone. Orig. art. has: 7 figures and 2 tables.

ASSOCIATION: Institut geokhimli i analiticheskoy khimii AN SSSR (Institute of Geochemistry and Analytical Chemistry, AN SSSR)

SUBMITTED: 310ct63

DATE ACQ: 30Apr64

ENCL: 00

SUB CODE: IC, OC

NO REF SOV: 008

OTHER: 003

Card 2/2

MINACHEV, Kh.M.; MARKOV, M.A.; BOGOMOLOV, V.I.; ENGLINA, F.E.

Transformation of cyclic alcohols on neodymium oxide. Izv.AN SSSR. Ser.khim. nc.l:13-17 Ja '64. (MIRA 17:4)

1. Institut organicheskoy khimii im. N.D.Zelinskogo AN SSSR.

DERBENTSEV, Yu.I.; MARKOV, M.A.; ISAGULYANTS, G.V.; MINACHEV, Kh.M.; BALANDIN, A.A., akademik; Prinimala uchastiye SHCHUKINA, O.K.

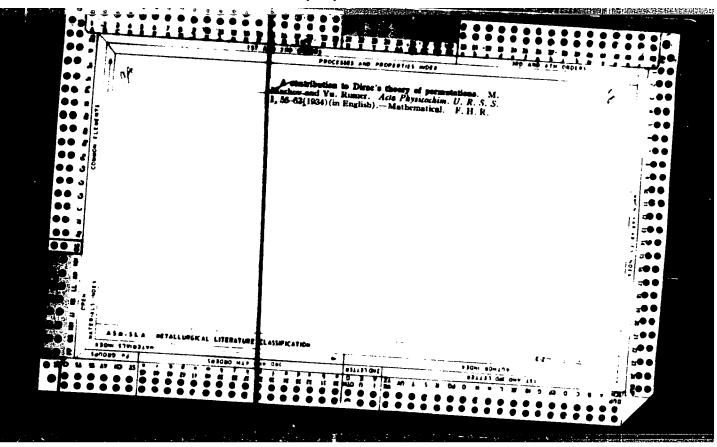
Mechanism of cyclohexane dehydrogenation over holmium oxide studied with the use of radiocarbon C¹⁴. Dokl. AN SSSR 155 no.1:128-131 Mr '64. (MIRA 17:4)

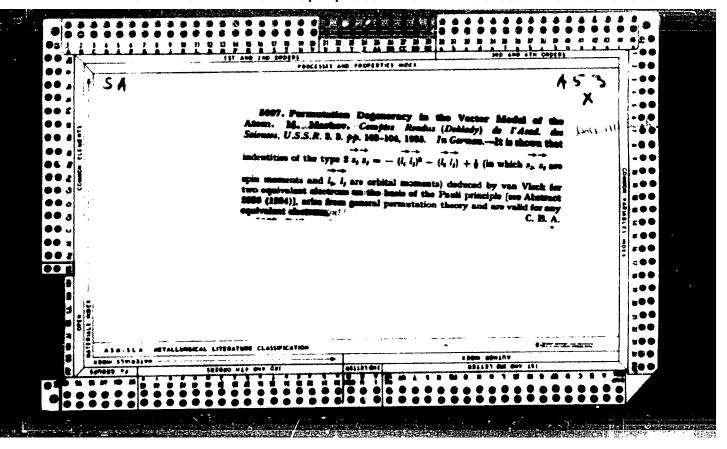
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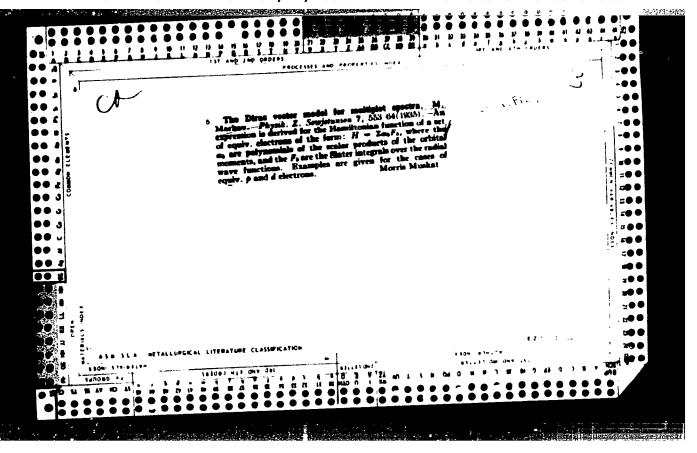
MINACHEV, Kh.M.; MARKOV, M.A.

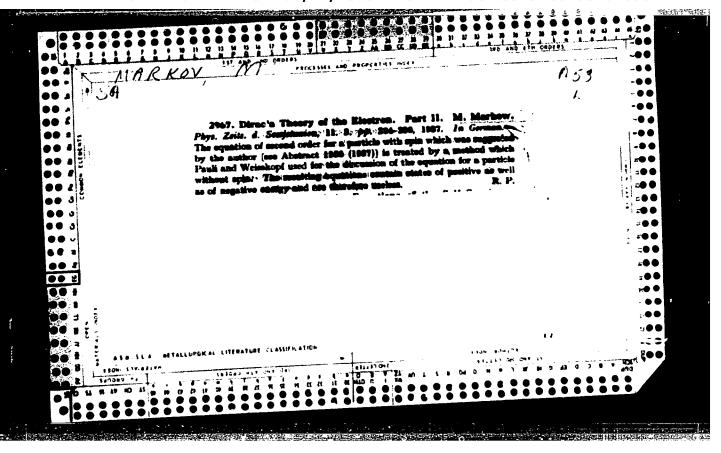
Dehydrogenation of cyclohexane on rare-earth oxides deposited on high-ash coals. Izv. AN SSSR. Ser. khim. no.9:1680-1682 '65. (MIRA 18:9)

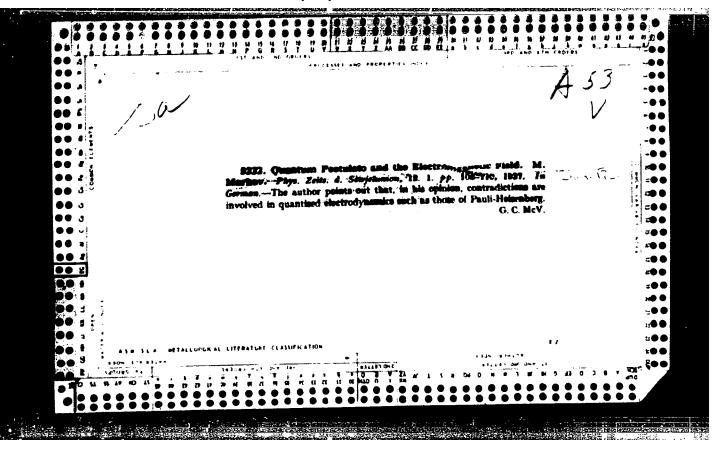
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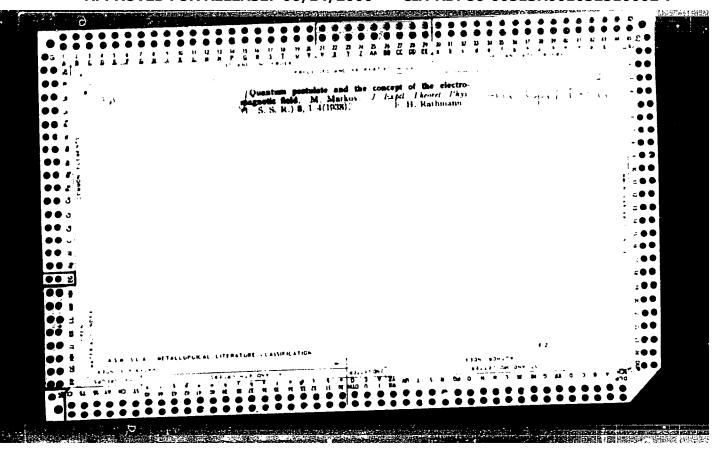


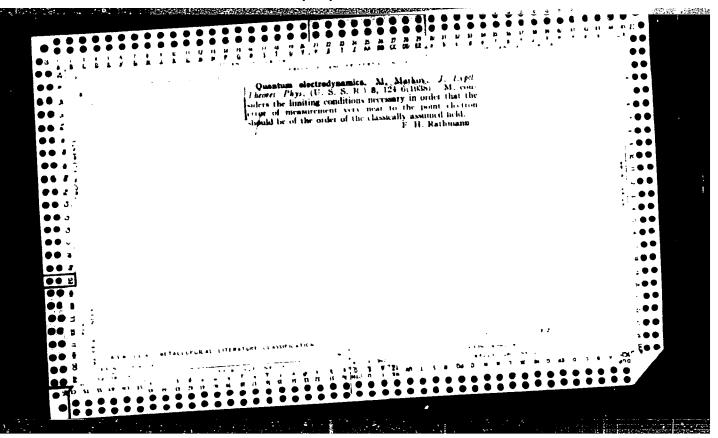


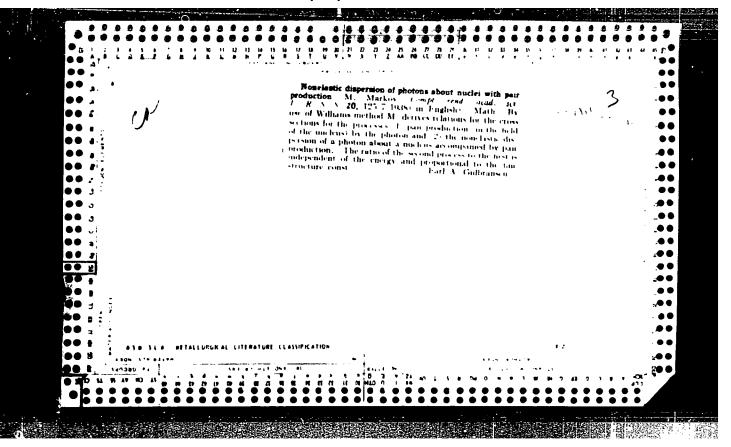












MARKOV, M. A.

"Role of the Zero-State of Rudiation Oscillators in the Higher Approximations of Quantum Electrodynamics," Dok.AR, 23, No. 9, 1939. Inst. Physics. cai. 201., 11 %.

"Concerning the Four Dimensional-Expanded Electron Relitivistic Quantum Jone"

Zhur. Phys., 453, No. 6, Vol. II, 1940

"On the Relativistically Invadiant (Dutain-Of Factor) In Electrodynamics," Dok.Ad., 16, No. 1, 1943. P.M. Lebedev Phys. Inst., Acad. Sci. c1943.

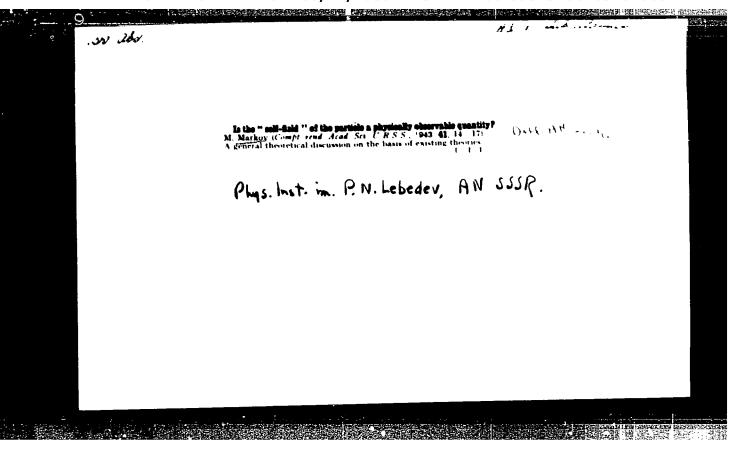
"Electrodynamics of Dirac-Fock-Podolsky and the Self-energy of the Electron," Dok.Ad., 16, No. 6, 1943. P.M. Lebedev Phys. Inst, Acad. Sci. c1943.

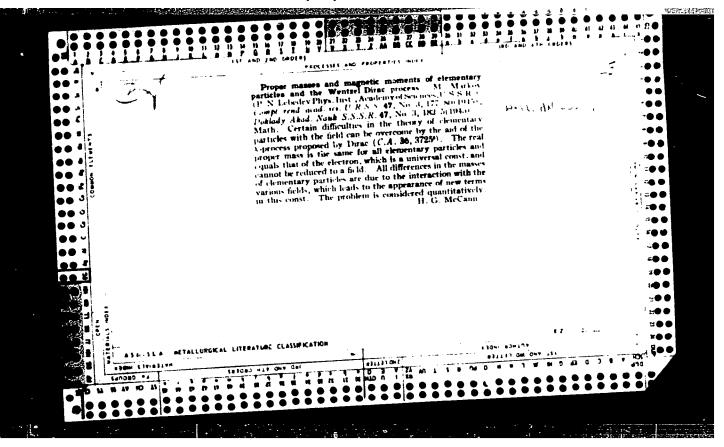
"The Many Body Problem in the Classical Relativistic Theory", Zhur. Phys., 12, 10. 1.

Vol. VII, 1943

¿. N. Lebedev Physical Institute, Academy of Science of the USSM, c1941.

"On the Criterion of Relativistic Invariance," Zhur. Eksper. i Teorge. Piz., 16, No. 9, 1948. Lebedev Physical Inst., Acad. Sci. USSM, -1946.





MARKOV, M.

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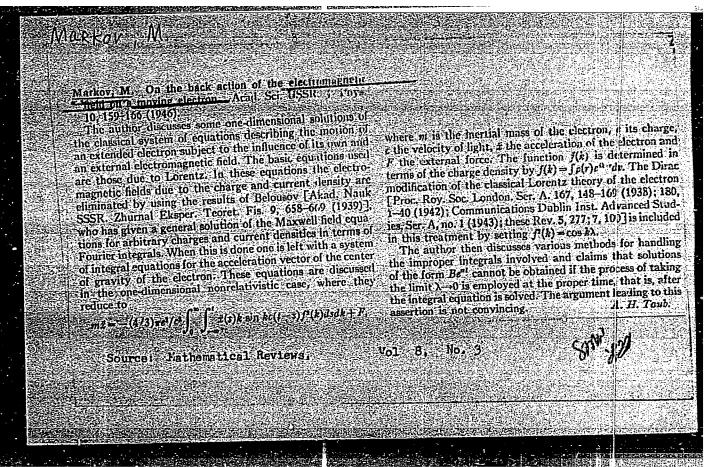
USER/Physics Relativity Quantum Mechanics Jul/Aug 1946

"A Certain Criterion of Relativistic Invariance," M. Markov, Lebedev Phys Inst, Acad Sci USER, 8 pp

"Journal of Physics USER" Vol I, No 4

Attempt to remove field theory difficulties by use of Hamilton-Jacobi equation, quantum equations of motion, Dirac's many-time formalism and Heitler-Peng "calculation scheme," considered from viewpoint of criterion of relativistic invariance. Received, 19 Aug 1945.

54291



Markoy M. On a Certain effection of relativistic never-ance. Acad Sci. USSRs 1; Phys. 10, 333-240 (1946)... If two points in space time are such that the vector joining them as space-like then the measurement of one physical quantity M at one of these points cannot react on the measurement of a quantity N at the second of these points. That is, the operators representing these quantities at the respective points must commute. The author applies this principle of "causal independence" to the operators representing the vector potential of an electromagnetic field and the Dirac-Hamiltonian for a system of n electrons interacting with this field. He uses the Dirac-Fock-Podolsky many time formalism to obtain a condition on the factors introduced by any "cut-off" process for the Fourier expan-sion of the vector potential. Introducing such cut-off factors is equivalent to modifying the commutation equations for the Pointer components of the vector potential. When these are written as $[A_{p_i}^+(k_j), A_{p_i}^-(k_j)] = \varepsilon(k_j, \lambda) \varepsilon_{p_i}$, instead of $[A_{p_i}^+(k_j), A_{p_i}^-(k_j)] = \delta_{p_i}$, the function $\varepsilon(k_i, \lambda)$ cannot be arbitrary but must satisfy the condition referred to above.

Source: Fathematical Reviews.

The muthor ahows that, if come = 1, 2, 3, 4, this condition is violated and hence the principle of causal micbendence at two relatively space-like points does not hold. He interprets the result as follows. The introduction of the function at a in the commutation relations is equivalent to postulating a certain charge distribution for the electron, as may be seen from an application of the many time ;oranalism. Signals are propagated across this charge distribution at a velocity greater than that of light and as a result the causal independence principle is violated. He then states that any choice of an integrable function for $\epsilon(k, \lambda)$ would determine a charge distribution and the same difficulty would then arise.

The author then reviews the singular function $c(k,\lambda)$ introduced in the Wentzel-Dirac limiting process and a generalization of this function. He also reviews the Heitler-Peng "calculation scheme" [Proc. Cambridge Philos. Soc. 38, 296-312 (1942)] in light of the causal independence principle and finds that this principle is satisfied.

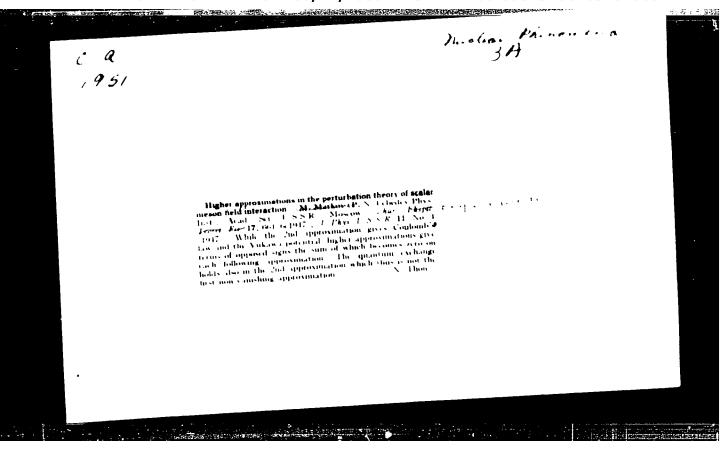
A.H. Taub (Stattle, Wash.)

8. No. 5

Diarkov, M. A. On the criterion of relativistic invariance. Udom of the system in question. So far only the limiting case Akad. Nauk SSSR. Zhuroal Ekaper, Teorot, Fiz. 16, 700. A=10 has been assumed to be physically significant. In the [Reviewer's note: the Russian title should be translated last section the author briefly discusses the "computational scheme" of Heitler and Peng [Proc. Cambridge Philos. Soc. 38, 296-312 (1942)] and Heisenberg's S-matrix [Z. Physik On a criterion of relativistic invariance."] Any relativistic theory based on the notion of space-time continuum ought to satisfy the following condition. (C) If (r_1, t_1) and (r_2, t_2) 120, 513-538 (1943); these Rev. 4, 292]. A fuller discussion are two world points which cannot be connected by a light in Acad. signal, i.e., if the inequality $c[h-L] < |r_1-r_2|$ holds, any 8, 303. I two events at these world points must not influence one is reserved for a later publication. [A translation appeared in Acad. Sci. USSR_J. Phys. 10, 333-340 (1946); these Rev. another. The formal relativistic invariance of a theory does V. Bargmann (Princeton, N. J.). not necessarily imply the condition (C). Using Dirac's many-time formalism the author investigates, from this point of view, several possible forms of quantum electro-dynamics (partly generalizing those which have appeared in the literature), in order to overcome the well-known convergence difficulties, in most of these theories a four vector k is introduced, which is then made to tend to zero. The author shows that as long as $\lambda \ge 0$ these theories do not satisfy the condition (C). LR eviewer's note. Since it is impossible to select a nonvanishing four vector in an invariant fashion, these theories cannot be considered relativis-tically invariant for λ=0 unless the components of λ arci introduced as variables which describe new degrees of free Source: Mathematical Reviews,

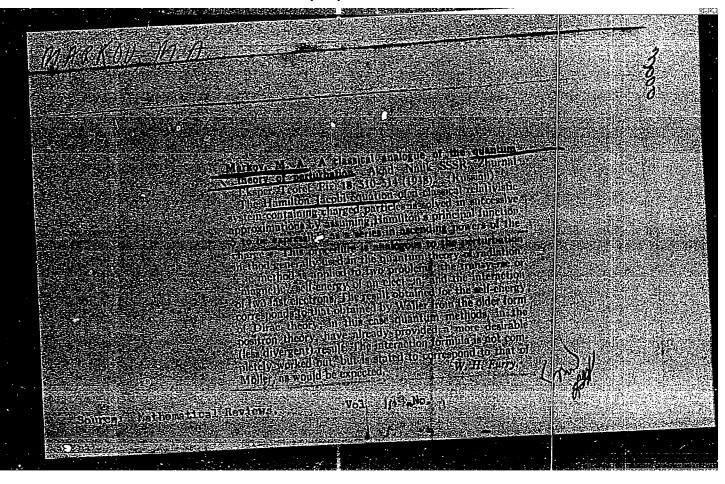
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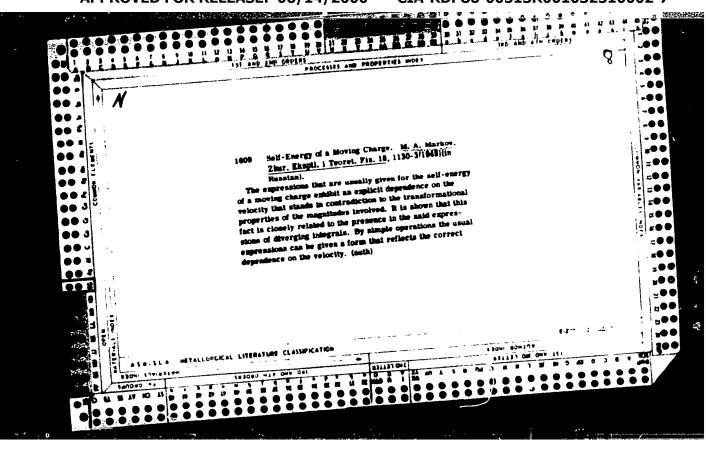
A SET MAKE	ov Menter and the second		7
	Markov, M. On the back action of field on a moving electron. Akad. Eksper Teoret 1/2 16, 800-8	the electromagnetic Nauk SSSI: Zhurni ID (1946), (Russan)	
3 Sour	Ekspe English summary) An English translation is reviewed		



MARKOV, M. Letter to the editor on the limiting λ -process. 2mr.eksp.i teor.fiz. 17 (MLRA 6:7) no.9:848 '47. 1. Fizicheskiy institut im. P.H. Lebedeva Akademii Hauk SSSR. (Quantum theory)

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CONTROL OF THE PROPERTY OF THE

MARKOV, M. A.

PA 19/49T109

USER/Smolear Physics - Endioactivity Oct 48
Factor Physics - Hostrino

"Theory of the Beta Decomposition in the Case of a Low Upper Boundary for the Beta-Spectrum and the Beutrino Mass," M. A. Markov, Phys Instimeni P. W. Lebedev, Acad Sci USER, 4 pp

"Zhur Eksper i Teoret Fiz" Vol XVIII, No 10

Examines beta decomposition assuming that neutrino mass is not zero. In cases of low upper beta-spectrum boundary, mass of neutrino plays essentially different role in variants of beta-decomposition theory. Results are illustrated using E³ as example. Submitted 20 Apr 48.

19/97109

MARKOV, M.

USSR/Nuclear Physics - Electrons, Pair

Feb 49

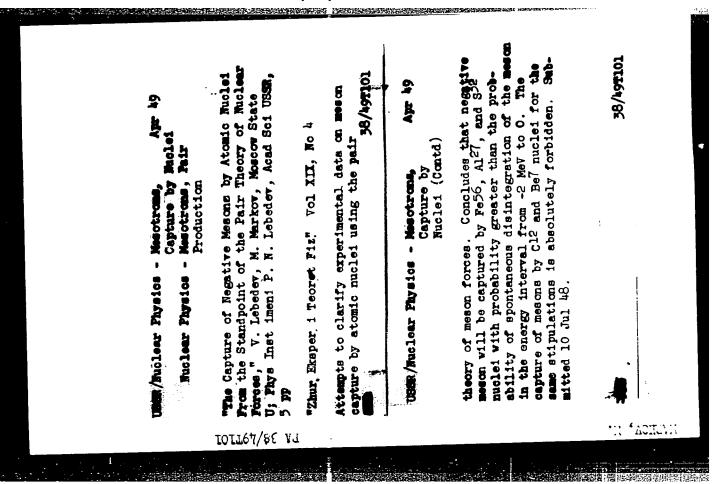
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Nuclear Physics - Neutrettos

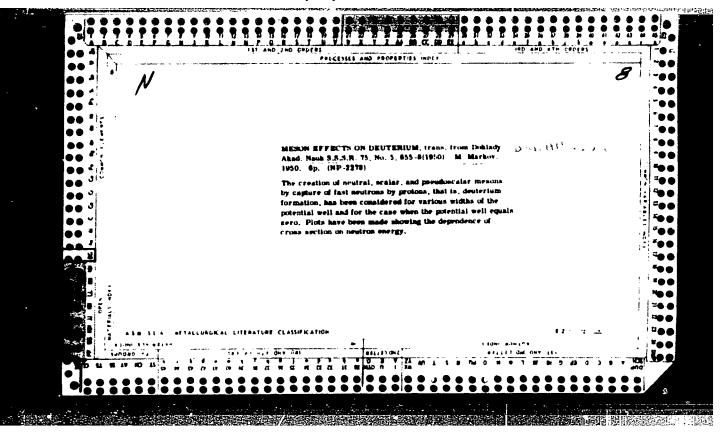
"The Pair Theory of Nuclear Forces," . Lomsadze, M. Markov, Moscow State U, Phys Inst imeni P. N. Lebedev, Acad Sci USSR, 3 pp

"Zhur Eksper i Teoret Fiz" Vol XIX, No 2

Shows that the neutretto and antineutretto lead to different laws of reaction in the theory of pair nuclear forces (spin of neutretto $\frac{1}{2}$)

PA32/49T76



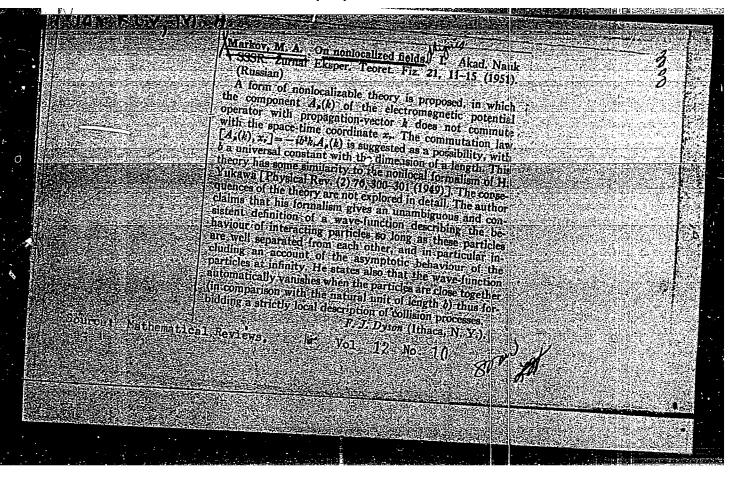


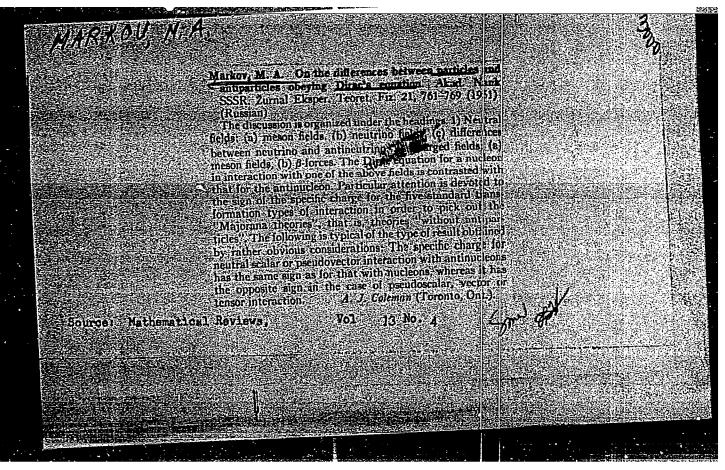
MARKOV, M. A., DRABNIKA, S. I., SUVOROV, S. G. (Editor), AKHLAMOV, S. N. (Tech. Editor), SOKOLOV, A. A. and FEYNBERG, Ye. A.

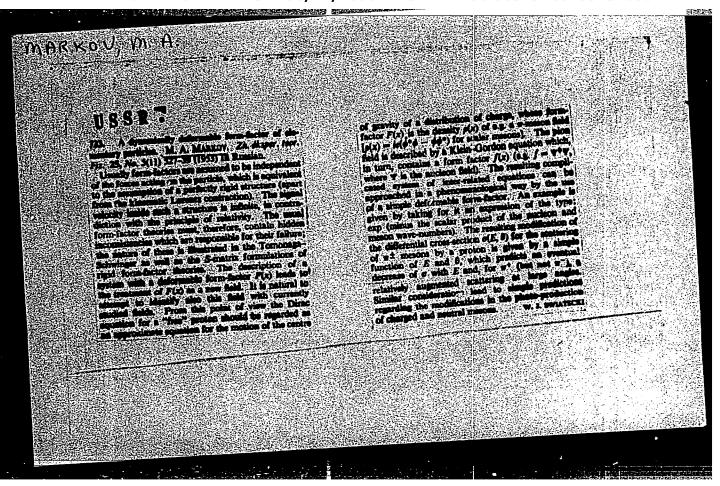
D. I. Blokhintsev, "Fundamentals of Quantum Mechanics", Osnovy Kvantovoy Mekhaniki, State Fress for Pechnical-Theoretical Literature.

Table of Contents W-17671, 5 Apr 1951

"APPROVED FOR RELEASE: 06/14/2000 CIA-RDP86-00513R001032510002-7







Bov 53

MARKOV, N. A.

USSR/Muclear Physics - Particle Theory

"Monlocal Fields and the Complex Nature of the 'Elementary' Particles (the Dynamically Deformable Form-Factor), M. A. Markov

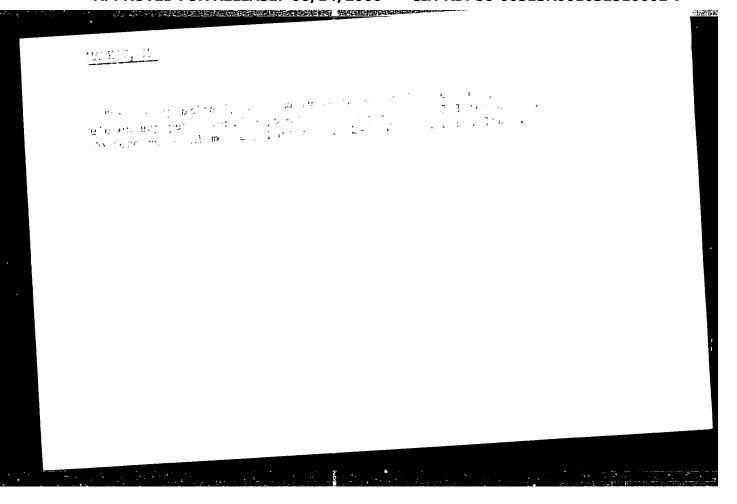
Usp Fiz Nauk, Vol 51, No 3, pp 317-341

Reviews existing attempts, Western and Soviet, to consider the elementary particles as fields (extensions). Discusses the phenomena appearing in the case of the approximate applicability of nondeformable form-factors (the creation of charged mesons by photons, and the

272T53

photo-generation of neutral mesons). Cites 8 Western and 3 Soviet references (D. Blokhintsev, Zherr 18, 1948; M. Karkov, Zherr, 10, 1940, and 21, 1951).

CIA-RDP86-00513R001032510002-7" APPROVED FOR RELEASE: 06/14/2000



MARKOV, M. A.

"On the Classification of Fundamental Particles," published by the Acaiemy of Eviences
"ISSE, 1955

English translation available Franch #5 E-3672

Paper presented at Rochester, N.Y. Conference on High Energy Nuclear Physics, 1955

UESR/Nuclear Physics - Hyperions and mesons

FD-2354

Card 1/2

Pub. 146 - 19/34

Author

: Markov, M., and Stakhanov, V.

Title

: Possible beta-decay of hyperions and k-mesons

Periodical

: Zhur. eksp. i teor. fiz. 28, /40, Jun 1955

Abstract

The writers of the present note consider as an obvious established fact that Λ^{0} -particles can enter the composition of complex nuclei on par with nucleons. Unknown, however, is how far the kinsh-i of hyperions and nucleons extends (V. Stakhanov, Diploma work, Moscow State University, 1954), in every case there being certain grounds for considering the N-particle as a nucleon situated in t certain excited state with all the consequences from this notion (V. Stakhanov, ibidem); e. g. such a nucleon could be beta-active: ^ →P+e +v. The purpose of the present note is to turn attention. to the fact that in this case thanks to the high upper bound of the energy of decay one can expect small lifetimes (tau sub-best) comparable with the observed lifetimes of Λ^{o} -particles, i. e. sub-pi $(\Lambda^0 \rightarrow \pi^- + P)$. The authors note that these ideas can be

Card 2/2 FD-2354

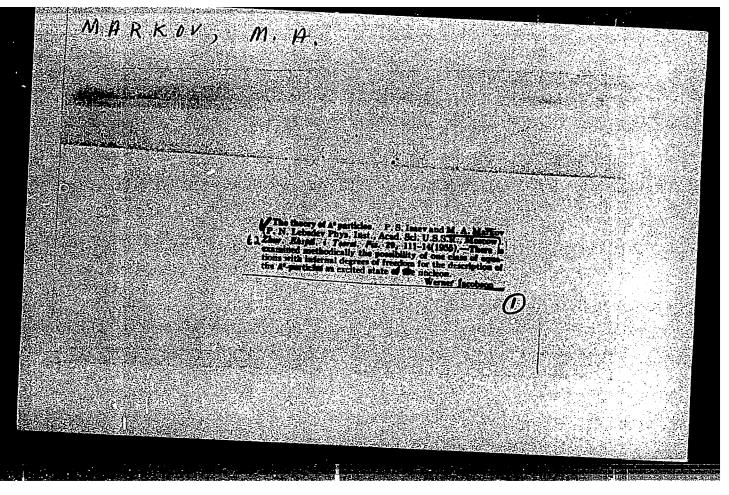
applied also to the case of K-mesons (M. A. Markov, Dok. Akai.

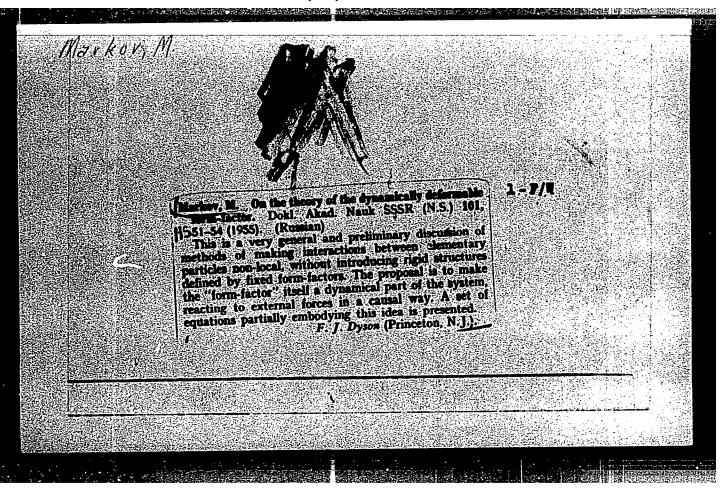
Nauk SSR, 101, 449, 1955) under certain conditions (zero spin, zeroay with emission of Fermi particles, etc.). Three references:

e. g. W. Fry and M. Swami, Phys. Rev. 96, 1954.

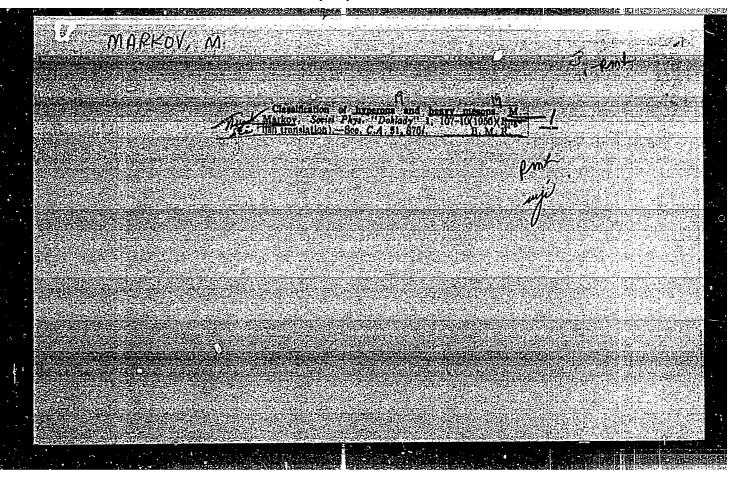
Institution : Physical Institute im. Lebedev, Acad. Sci. USSR

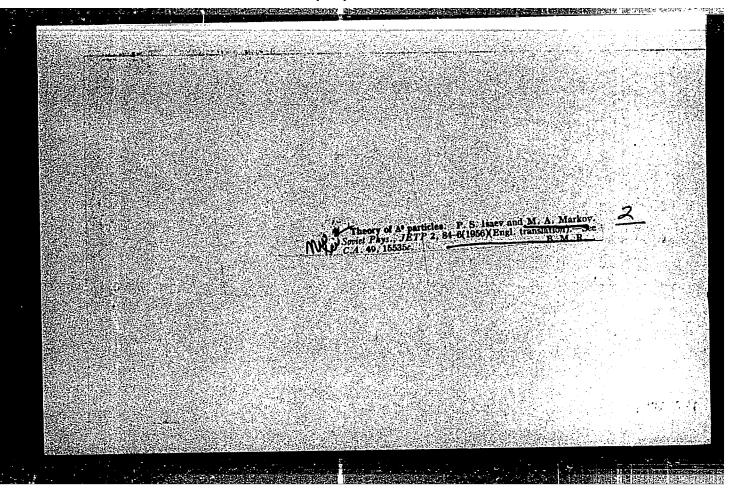
Submitted : Feb 9, 1955





Physics Pub. 22 - 14/49 Markov, M.A., Member-Correspondent of the Acad. of Sc., USSR 111 "Hyperons", i.e., muclons in an excited state, as a possible cause (mechanism) of mass particle formation Dok. AN 8858 101/3, 449-452, Mar 21, 1955 An attempt is made to connect (theoretically) the mass formation of particles (nuclons and anti-nuclons) with the disintegration of "hyperons", 1.0 . nuclons which are excited by -mesons and are in the cascade form of disintegration. A number of peculiarities which may be met in the theories dealing with such subjects are pointed out (a few exemplary cases are discussed). Two references: 1 USSR and 1 USA (1949-1955). The Acad. of Sc., USSR, P. N. Lebedev Physical Institute Institution 1 December 21, 1954 is tod





Markov, M.
USSR/Theoretical Physics - Quantum Field Theory

B-6

Abst Journal: Referat Zhur - Fizika, No 12, 1956, 33815

'Author : Markov, M.

Institution: Physics Institute, Academy of Sciences USSR

Title: Remark Concerning the Systematization of Hyperons in Heavy Mesons

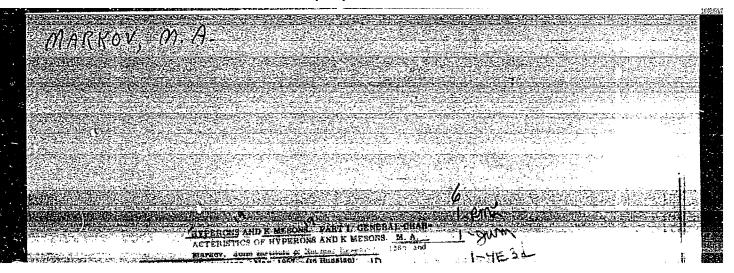
Original

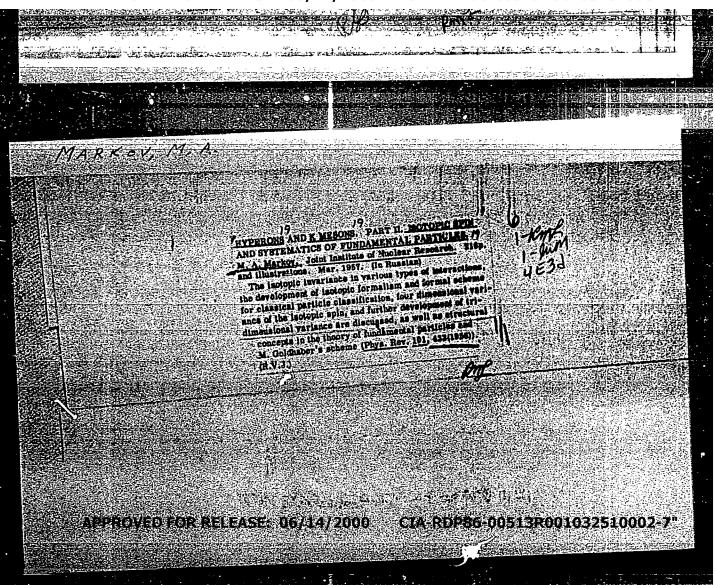
Periodical: Dokl. AN SSSR, 1956, 106, No 5, 814-817

Abstract: A systematization of hyperons and K-mesons is proposed. The hyperons are considered as excited states of nucleons, whereby the number of the excitation coincides with the value of the oddity of the given hyperon. Generalizing the Fermi-Yang hypothesis, the author considers K-mesons as bound systems, consisting of a nucleon (antinucleon) and an antihyperon (hyperon). The suggested systematiza-

tion is compared with the Gell-Mann systematization.

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MARKON, M

AUTHOR:

MARKOV,M.

TITLE:

Comments to D. KONOSENKO'S Work on "Semi-conductor Bolometers"

(Po povodu stati D.Konosenko: Poluprovodnikovye bolometry, Russian) Uspekhi Fiz. Naua, 1957, Vol 62, Nr 2, pp 197-199 (U.S.S.R.)

ABSTRACT:

PERIODICAL:

In his paper KCNOSENKO endeavors to systemize the bolometers made of semiconductors and to establish their importance for the measuring of infrared radiation as compared with other bolometers. According to MARKOV this statement contains a number of errors

53-2-9/9

which he mentions and discusses:

KONOSENKO, when judging the sensitivity of bolometers, ignores the particular features of the electric amplifiers connected to the bolometer and therefore his evaluation of them is wrong.

KONOSENKO'S opinion that semiconductor bolometers are better

than metal bolometers cannot be shared.

MARKOV proves to KUNCSENKO that he did not display enough care when citing published works as references. Furthermore, MARKOV declares several of KONOSENKO'S statements concerning various properties as absolutely wrong. KONOSENKO'S paper appeared in Uspekhi

Fizicheskikh Nauk 1955, Vol. 56, Nr 2, p.283.

PRESENTED BY:

SUBMITTED: AVAILABLE:

Library of Congress

Card 1/1

21(8)

PHASE I BOOK EXPLOITATION

sov/1413

Markov, Moisey Aleksandrovich

Giperony i K-mezony (Hyperons and K-Mesons) Moscow, Fiznatgiz, 1958. 343 p. (Series: Sovremennyye problemy fiziki) 5,000 copies printed.

Eds.: Alekseyev, D.M. and I.I. Denisov; Tech. Ed.: Akhlsmov, S.N.

PURPOSE: This book is intended for theoretical physicists and scientists in the field of nuclear physics and the physics of elementary particles. It may also be used by students of advanced courses and Aspirants specializing in nuclear physics.

COVERAGE: A detailed review is given of experimental and theoretical work on the physics of two cosmic ray particles: hyperons and K-mesons.

TABLE OF CONTENTS

Foreword

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PART I. GENERAL CHARACTERISTICS OF HYPERONS AND K-MESOMS

Card 1/6

CIA-RDP86-00513R001032510002-7" APPROVED FOR RELEASE: 06/14/2000

MARKOV, M. A.

"An Example of a Field Theory with Indefinite Metric in Hilbert Space."

The Example of a Field Theory with Indefinite Metric in Hilbert Space."

The Example of a Field Theory with Indefinite Metric in Hilbert Space."

The Example of a Field Theory with Indefinite Metric in Hilbert Space."

Joint Inst. of Muclear Research, Dubna, USSR

A field theory is considered in which the Green's functions have singularities not on a cone but on a hyperboloid, The Metric of the corresponding Hilbert space is indefinite. Equations of motion for the boson and fermion fields are derived. The theory goes over to the usual field theory when a certain constant(characteristic length) approaches zero. The ultimate aim of the investigation is an attempt to find a possible reasonable interpretation of the negative energy density of the boson fields, j of the negative probabilities, and of other similar features which appear in the theory, by appropriately modifying the concept of a particle coordinate so as to obtain the proposed propagation functions.